

Modified zirconium-based metal organic framework by 4-aminobenzoic acid to enhance CO₂ adsorption performance

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Zr-based MOF (UiO-66) is known as a leading candidate of adsorbents applied for gas separation process due to high surface area, easily tuned porous properties; chemical surface and diversity in synthesis method. In compared to other MOF-type, however, UiO-66 shows CO₂ capture capacity modestly. Therefore, UiO-66 material should be modified its characterization to improve CO₂ adsorption capacity. In this work, modified UiO-66 was prepared in one step synthesis between ZrCl₄ and linker mixed by terephthalic acid and 4-aminobenzoic acid (ABA) systematically. The presence of ABA in reaction system induced missing-linker defects, simultaneously appearance of NH₂ groups in framework, causing to improve CO₂ capture capacity. The obtained materials were analysed by SEM, XRD, FTIR, XPS and TGA. The adsorption isotherms of CO₂ and N₂ on as-prepared adsorbents were recorded at 298 K. The results showed that UiO-66#10-NH₂ showed a CO₂ uptake amount of 2.1 mmol/g, which is higher than that of UiO-66 about 45%. Also, the CO₂/N₂ selectivities based on ideal adsorbed solution theory (IAST) and isosteric heat of adsorption were investigated to evaluate gas separation performance of adsorbents