

## A Fundamental Understanding of Carbon Dioxide Capture by Magnesium Oxide-Based Sorbents at High Temperature

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We successfully carried out the direct observation of the triple-phase capture process involved in the sorption of gaseous CO<sub>2</sub> over molten EM-promoted solid MgO via in situ TEM and detected the changes happening within the system as a whole in real time. We also conducted studies on the interface between EM over solid MgO and MgCO<sub>3</sub> formed during sorption, which further clarifies the interaction between MgO and EM. Secondly, we report the feasibility of using commercially available hydrotalcites as stabilizing additives for NaNO<sub>3</sub>-promoted MgO sorbents. We demonstrate how the simple addition of stabilizers such as hydrotalcites which interact with MgO and NaNO<sub>3</sub> can alter and improve the MgCO<sub>3</sub> formation on molten nitrate-promoted sorbents, resulting to improved stability and increased cumulative CO<sub>2</sub> capture capacity. Lastly, we report simple methodology to identify the Mg ion inversion in the MgO-Al<sub>2</sub>O<sub>3</sub> by solid state NMR, and verify the correlation with CO<sub>2</sub> capture capacity of MgO-rich MgO@MgO-Al<sub>2</sub>O<sub>3</sub> spinel structures.