Photoelectrochemical reduction of carbon dioxide on NiO/quantum dots heterostructure electrodes

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Recently, as the global warming is serious problem, the conversion of carbon dioxide to carbon monoxide or hydrocarbons which are the promising renewable chemical fuels is expected to solve this environmental problem. However, the product selectivity of CO2 conversion is not good compared to H2 generation. The study about electron-transfer kinetics is expected to be a major role for understanding the CO2 selectivity. Here, we used NiO/quantum dots heterostructure electrodes for photoelectrochemical (PEC) reduction of CO2 with various size of QDs and compared the product selectivity of CO2 reduction depending on the size of nanocrystal with electron-transfer kinetics.