## The effect of $SO_2$ and NO on $CO_2$ hydrate

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The captured  $\mathrm{CO}_2$  for  $\mathrm{CO}_2$  sequestration includes traces of impurities such as  $\mathrm{SO}_X$  and  $\mathrm{NO}_X$ . However, Only Beeskow-strauch reported equilibrium data for  $\mathrm{CO}_2$ -water- $\mathrm{SO}_2$  mixture and no literature data are known to present authors for  $\mathrm{CO}_2$ -water- $\mathrm{NO}$  mixture. In this study, the effect of  $\mathrm{SO}_2$  and  $\mathrm{NO}$  on  $\mathrm{CO}_2$  hydrate formation in the range of 1 to 10% impurities of the mixed gas determined in aqueous-hydrate-vapor equilibria region. The isobaric hydrate dissociation temperatures of formed hydrate were measured at the constant pressure from 1.5 to 3.5 MPa. Duhem's theorem was thermodynamically satisfied by measuring temperature, pressure and overall composition of all components. Present data were compared with literature data.