Selective Catalytic Reduction of NOx over Supported MnOx Catalysts

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MnOx supported on Al_2O_3 , TiO_2 , ZrO_2 and SiO_2 were prepared by a precipitation method and a wet impregnation method, and their catalytic activity for the low-temperature selective catalytic reduction (SCR) of NOx with NH $_3$ in the presence of excess O_2 was examined. The activity of the catalysts prepared by a precipitation method was higher than that of catalysts prepared by an impregnation method. The activity follows in the order: $MnOx/TiO_2 \approx MnOx/Al_2O_3 > MnOx/SiO_2 > MnOx/ZrO_2$. Supported MnOx catalysts prepared by a precipitation method showed an amorphous manganese oxide phase and those prepared by an impregnation method exhibited a crystalline MnO_2 phase, respectively. This high de-NOx activity of MnOx supported catalyst prepared by a precipitation method at low temperature is probably due to their amorphous framework structure and the existence of residual carbon oxide species.