

TRS FT-IR Spectroscopy of Surface Intermediates of C₂H₄ Hydrogenation over Pt/Al₂O₃ Catalyst under Reaction Conditions

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Time-resolved rapid-scan FT-IR spectra of ethylene hydrogenation over alumina-supported Pt catalyst at 200 °C were recorded under continuous H₂/N₂ flow and pulsed release of C₂H₄ (30 millisecond duration). Two surface species were observed, namely ethylidyne (CH₃CPt₃) with peaks at 2880 and 1339 cm⁻¹ (lifetime 300 ± 50 msec), and a substantially shorter-lived intermediate with an intense band at 1200 cm⁻¹ and weak absorptions in the 2875-2860 cm⁻¹ region (lifetime around 100 msec). Comparison of the C₂H₄ + H₂ results with those of experiments using D₂ or C₂D₄ suggests that the 1200 cm⁻¹ species is a surface ethyl intermediate (CH₃CH₂Pt). This is the first observation on the lifetime of surface ethyl species under reaction conditions. The rise of the final ethane product, monitored by the ν(CH) absorption at 2893 cm⁻¹ was found to reach a maximum already in the first recorded time slice. This suggests that the observed CH₃CH₂Pt species is a surface-trapped form of the kinetically relevant, only weakly interacting C₂H₅ radical intermediate.