## **Dispersion Models**

### **Exposure to Release**

#### Predict effects of exposure near the surface.



#### **Stages**

- 1. Source
- 2. Acceleration, Diffusion
- 3. Gravity
- 4. Transition
- 5. Surface
- 6. Turbulence

#### Predict % affected by the exposure.

## **Fluids Beyond the Sources**

- **Effluent properties dominate near a leak**
- Material then migrates and mixes with air
- **Ambient conditions eventually dominate** 
  - Pressure, temperature, wind velocity, humidity, sun light
- Transport and mixing with air at a vapor cloud boundary
- Isopleth: constant concentration boundary of a vapor cloud

### **Accidental Flow**



## **Dispersion Modeling Needed**

Goals: prevent releases; mitigation
Prevent

Inherent safety practices: reduction, substitution, attenuation

Process design and integrity

**4**PSM management; PHA

Mitigation measures

Emergency response planning

### **Hazard Levels**

#### Concentration

### Air velocity and turbulence

# Time period of release; C(time) following release

Position of cloud relative to ground

### **Gaussian Dispersion Pattern**



### **Dispersion Parameters**

- Cloud of effluents expands, mixes with air
- Mixing dilutes the effluent: C decreases
- **4** Lower downwind  $C \Rightarrow$  greater area affected
- ♣ Dominant dispersion mechanism: turbulent dispersion ⇒ horizontal and vertical movement
- Mixing rate depends on *u*, atmosphere stability, buoyancy

#### Light winds, strong sun ⇒ most unstable: rapid diffusion



#### **Plume & Puff**







## **Plume Model**

- Steady state concentration from a continuous source, e.g. smokestack
- Initially increases in size, additions from source
- Steady state: same amount of effluent added to plume as is mixed with air; constant volume
- Source stopped: plume size decreases as mixing with air is dominant, plume returns to source origin, finally disappears.

### Puff Model

- Cloud formed from a fixed amount of effluent, e.g., from a ruptured vessel
- Release over a short period of time that source is active
- Movement from source: dependent on air velocity
- Material mixes with air, boundary diminished in size, finally disappears

$$\frac{\partial \langle C \rangle}{\partial t} + \langle u_j \rangle \frac{\partial \langle C \rangle}{\partial x_j} = \frac{\partial}{\partial x_j} \left( K_j \frac{\partial \langle C \rangle}{\partial x_j} \right)$$
  
Plume  
Plume  
Of Puff  
1. SS,  $\langle u_j \rangle = 0, K_j = K^*$   
3. NSS,  $\langle u_j \rangle = 0, K_j = K^*$   
4. SS,  $\langle u_j \rangle = \langle u_x \rangle = u, K_j = K^*$   
6. SS,  $\langle u_j \rangle = \langle u_x \rangle = u, K_x, K_y, K_z$   
9. SS,  $\langle u_j \rangle = \langle u_x \rangle = u, K_x, K_y, K_z$   
 $\frac{\partial \langle C \rangle}{\partial x_j} = \frac{\partial}{\partial x_j} \left( K_j \frac{\partial \langle C \rangle}{\partial x_j} \right)$   
Puff  
Puff  
Puff  
Puff  
Puff  
 $\langle C \rangle(r) = \frac{Q_m}{4\pi K^* r} erfc \left( \frac{r}{2\sqrt{K^* t}} \right)$   
 $\langle C(r) \rangle = \frac{Q_m}{4\pi K^* r} exp \left( -\frac{u(r-x)}{2K^*} \right)$ 

9. 
$$SS_{10/15/20j} > = < u_x > = u, K_x, K_y, K_H Source$$

8. 
$$< u_j >= 0, K_x, K_y, K_z$$
 Source

## **Neutrally Buoyant Dispersion**

No reactions; small effect of molecular diffusion

- **4** Mixing mechanism: air turbulence
- **4** Turbulence  $\Rightarrow$  fluctuations in *C*, *u*

$$\frac{\partial C}{\partial t} + \frac{\partial}{\partial x_j} (u_j C) = 0$$

$$u_j, \text{ air velocity}$$

$$C, \text{ concentration}$$

$$u_j = \left\langle u_j \right\rangle + u'_j \longrightarrow C = \left\langle C \right\rangle + C'$$

 $u'_j, C'$ , fluctuation components

## Eddy Diffusivity, *K*<sub>j</sub>

**Represent C fluctuation due to turbulence** 

$$\left\langle u_{j}^{\prime}\right\rangle = 0; \left\langle C_{j}^{\prime}\right\rangle = 0$$
  
 $\left\langle u_{j}^{\prime}C^{\prime}\right\rangle = -K_{j}\frac{\partial\langle C\rangle}{\partial x_{j}}$ 

#### **Governing equation:**

$$\frac{\partial \langle C \rangle}{\partial t} + \left\langle u_j \right\rangle \frac{\partial \langle C \rangle}{\partial x_j} = \frac{\partial}{\partial x_j} \left( K_j \frac{\partial \langle C \rangle}{\partial x_j} \right)$$

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### 1. Steady State, Point Release, No Wind

↓ Q<sub>m</sub> constant; C independent of t, wind, u ~ 0
 ↓ Constant K<sub>i</sub> = K\* in all directions



Polar coordinates, integrate over *r* :

$$\langle C \rangle (r) = \frac{Q_m}{4\pi K^* r}$$

$$r = \sqrt{x^2 + y^2 + z^2}$$

### 2. Puff Release, No Wind

↓ Wind velocity, *u* ~ 0
 ↓ Constant *K<sub>i</sub>* = *K*\* in all directions

$$\frac{1}{K^*}\frac{\partial \langle C \rangle}{\partial t} = \frac{\partial^2 \langle C \rangle}{\partial x^2} + \frac{\partial^2 \langle C \rangle}{\partial y^2} + \frac{\partial^2 \langle C \rangle}{\partial z^2}$$

Instantaneous concentration:



#### 3. Non SS Point Release, No Wind

↓ Q<sub>m</sub> constant; wind, u ~ 0
 ↓ Constant K<sub>i</sub> = K\* in all directions

$$\frac{1}{K^*}\frac{\partial \langle C \rangle}{\partial t} = \frac{\partial^2 \langle C \rangle}{\partial x^2} + \frac{\partial^2 \langle C \rangle}{\partial y^2} + \frac{\partial^2 \langle C \rangle}{\partial z^2}$$

Integrate instantaneous concentration:

$$\langle C(r,t) \rangle = \frac{Q_m}{4\pi K^* r} \operatorname{erfc}\left(\frac{r}{2\sqrt{K^* t}}\right)$$

#### **Error function & its Integration**



METU-NCC

### 4. SS Point Source with Wind

*Q<sub>m</sub>* constant; *C* independent of *t* Wind in *x* direction, *u<sub>x</sub>* constant
 Constant *K<sub>j</sub>* = *K*\* in all directions



Centerline for  $y^2 + z^2 \ll x^2$ :  $\langle C(x) \rangle = \frac{\mathcal{Q}_m}{\mathcal{Q}_m}$ 

### 5. Puff with No Wind, K<sub>j</sub> Varies

- ↓ Q\*<sub>m</sub> constant; Puff release
  ↓ No wind(<u<sub>i</sub>> =0)
- $K_j \neq K^*$ , but constant in all directions

$$\frac{\partial \langle C \rangle}{\partial t} = K_x \frac{\partial^2 \langle C \rangle}{\partial x^2} + K_y \frac{\partial^2 \langle C \rangle}{\partial y^2} + K_z \frac{\partial^2 \langle C \rangle}{\partial z^2}$$

$$\langle C \rangle(x,y,z,t) = \frac{Q_m^*}{8(\pi t)^{3/2} \sqrt{K_x K_y K_z}} \exp \left[ -\frac{1}{4t} \left( \frac{x^2}{K_x} + \frac{y^2}{K_y} + \frac{z^2}{K_z} \right) \right]$$

#### 6. SS Point Source with Wind, K<sub>i</sub> Varies

↓ Q<sub>m</sub> constant; C independent of t
 ↓ Wind in x direction, u<sub>x</sub> constant
 ↓ K<sub>i</sub> ≠ K\*, but constant in all directions



#### 7. Puff with Wind

#### Q<sup>\*</sup><sub>m</sub> constant; Puff release

- **4** Wind in x direction only( $\langle u_j \rangle = \langle u_x \rangle = u = \text{constant}$ )
- $K_j \neq K^*$ , but constant in all directions

$$\frac{\partial \langle C \rangle}{\partial t} = K_x \frac{\partial^2 \langle C \rangle}{\partial (x - ut)^2} + K_y \frac{\partial^2 \langle C \rangle}{\partial y^2} + K_z \frac{\partial^2 \langle C \rangle}{\partial z^2}$$
$$C \rangle (x, y, z, t) = \frac{Q_m^*}{8(\pi t)^{3/2} \sqrt{K_x K_y K_z}} \exp\left[-\frac{1}{4t} \left(\frac{(x - ut)^2}{K_x} + \frac{y^2}{K_y} + \frac{z^2}{K_z}\right)\right]$$

#### 8. Puff with No Wind, Source on Ground

↓ Q\*<sub>m</sub> constant; Puff release
 ↓ No wind(<u<sub>j</sub>> =0)
 ↓ K<sub>i</sub> ≠ K\*, but constant in all directions

$$\frac{\partial \langle C \rangle}{\partial t} = K_x \frac{\partial^2 \langle C \rangle}{\partial x^2} + K_y \frac{\partial^2 \langle C \rangle}{\partial y^2} + K_z \frac{\partial^2 \langle C \rangle}{\partial z^2}$$

$$\langle C \rangle(x,y,z,t) = \frac{Q_m^*}{(\pi t)^{3/2} \sqrt{K_x K_y K_z}} \exp\left[-\frac{1}{4t} \left(\frac{x^2}{K_x} + \frac{y^2}{K_y} + \frac{z^2}{K_z}\right)\right]$$

#### Impervious boundary

#### 9. SS Point Source with Source on Ground

↓ Q<sub>m</sub> constant; C independent of t
 ↓ Wind in x direction, u<sub>x</sub> constant
 ↓ K<sub>i</sub> ≠ K\*, but constant in all directions



#### 10. SS Point Source with Source at Height H<sub>r</sub> above the Ground

- $\mathbf{Q}_m$  constant; C independent of t
- **4** Wind in x direction,  $u_x$  constant
- $K_j \neq K^*$ , but constant in all directions



### **Consequence Analysis (Ex)**



#### **Risk contour**

### **Consequence Analysis (Ex)**



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## **Pasquill-Gifford Model**

K<sub>i</sub> values difficult to measure



- ▲ Define: dispersion coefficient ~ st dev for C  $\sigma_i^2 = \frac{1}{2} \langle C \rangle^2 (ut)^{2-n} \quad i = x, y, z; n, \text{ parameter}$
- $\sigma_j$ : functions of downwind distance, *x*, and atmospheric conditions in stability classes, *A* - *F*, based on sunlight and wind speed. Tab 5-1, p. 187

 $\sigma_j$  values for rural or urban plumes, or puffs from Figs 5-10 - 5-12 or Tabs 5-1 - 5-3, pp., 187-189

#### Nighttime conditions<sup>4</sup> Surface Thin overcast Daytime insolation<sup>3</sup> wind speed or >4/8<3/8 (m/s) Strong Moderate Slight low cloud cloudiness $\leq 2$ A-B $\mathbf{F}^{5}$ A B $F^{5}$ 2 - 3A-BB $\mathbf{C}$ $\mathbf{F}$ E. 3 - 4B-CCB $D^6$ Ε 4 - 6 $\mathbb{C}$ C-D $D^6$ D6 D٥ $\mathbf{C}$ >6D6 $D^6$ $D^6$ $D^{6}$

 Table 5-1
 Atmospheric Stability Classes for Use

 with the Pasquill-Gifford Dispersion Model <sup>1,2</sup>

- A: Extremely unstable B: Moderately unstable C: Slightly unstable
- **D: Neutrally stable**
- E: Slightly stable
- **C: Moderately stable**

**Table 5-2** Recommended Equations for Pasquill-Gifford Dispersion Coefficients for Plume Dispersion<sup>1,2</sup> (the downwind distance *x* has units of meters)

Pasquill-Gifford stability class	$\sigma_{y}$ (m)	$\sigma_{z}(m)$
Rural conditions		
Α	$0.22x(1 + 0.0001x)^{-1/2}$	0.20x
В	$0.16x(1 + 0.0001x)^{-1/2}$	0.12x
С	$0.11x(1 + 0.0001x)^{-1/2}$	$0.08x(1 + 0.0002x)^{-1/2}$
D	$0.08x(1+0.0001x)^{-1/2}$	$0.06x(1 + 0.0015x)^{-1/2}$
Е	$0.06x(1+0.0001x)^{-1/2}$	$0.03x(1 + 0.0003x)^{-1}$
F	$0.04x(1 + 0.0001x)^{-1/2}$	$0.016x(1 + 0.0003x)^{-1}$
Urban conditions		
A-B	$0.32x(1 + 0.0004x)^{-1/2}$	$0.24x(1 + 0.0001x)^{+1/2}$
D	$0.22x(1 + 0.0004x)^{-1/2}$	0.20x
D	$0.16x(1 + 0.0004x)^{-1/2}$	$0.14x(1 + 0.0003x)^{-1/2}$
E-F	$0.11x(1 + 0.0004x)^{-1/2}$	$0.08x(1+0.0015x)^{-1/2}$

**Table 5-3** Recommended Equations for Pasquill-Gifford Dispersion Coefficients for Puff Dispersion<sup>1,2</sup> (the downwind distance x has units of meters)

Pasquill-Gifford stability class	$\sigma_y$ (m) or $\sigma_x$ (m)	σ <sub>z</sub> (m)
A	$0.18x^{0.92}$	$0.60x^{0.75}$
В	$0.14x^{0.92}$	$0.53x^{0.73}$
С	$0.10x^{0.92}$	$0.34x^{0.71}$
D	$0.06x^{0.92}$	$0.15x^{0.70}$
E	$0.04x^{0.92}$	$0.10x^{0.65}$
F	$0.02x^{0.89}$	$0.05x^{0.61}$



Figure 5-10 Dispersion coefficients for Pasquill-Gifford plume model for rural releases.



Figure 5-11 Dispersion coefficients for Pasquill-Gifford plume model for urban releases.



Figure 5-12 Dispersion coefficients for Pasquill-Gifford puff model.

### **Atmospheric Stability Classes**



**Day Temperature** 

Stability classes classify level of turbulance: Asystematical Asystemati

### 11. Puff, Ground Source, *u* Constant

$$\left\langle C(x,y,z,t)\right\rangle = \frac{Q_m^*}{\sqrt{2}\pi^{3/2}\sigma_x\sigma_y\sigma_z} \exp\left\{-\frac{1}{2}\left[\left(\frac{x-ut}{\sigma_x}\right) + \frac{y^2}{\sigma_y} + \frac{z^2}{\sigma_z}\right]\right\}$$

#### Ground concentration: z = 0

Ground concentration along x: y = z = 0

#### Center of moving puff, x = ut:

$$\langle C(ut,0,0,t) \rangle = \frac{Q_m^*}{\sqrt{2}\pi^{3/2}\sigma_x \sigma_y \sigma_z}$$

### **Total Dose**

$$D_{tid}(x, y, z) = \int_0^\infty \langle C \rangle(x, y, z, t) dt$$

### Puff, ground source, constant *u*:

**Ground level:** 
$$D_{tid}(x,y,0) = \frac{Q_m^*}{\pi \sigma_y \sigma_z u} \exp\left(-\frac{1}{2}\frac{y^2}{\sigma_y^2}\right)$$

Along **x**: 
$$D_{tid}(x,0,0) = \frac{Q_m^*}{\pi \sigma_v \sigma_z u}$$

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### 12. Plume, Ground Source, u Constant

$$\langle C(x, y, z) \rangle = \frac{Q_m}{\pi \sigma_y \sigma_z u} \exp \left[ -\frac{1}{2} \left( \frac{y^2}{\sigma_y} + \frac{z^2}{\sigma_z} \right) \right]$$

Ground C(x,y,0) : z = 0

Ground, C(x,0,0) along x: y = z = 0

Isopleth concentration, *C*\* :

$$y = \pm \sigma_{y} \sqrt{2 \ln \left( \frac{\langle C(x,0,0,t) \rangle}{\langle C(x,y,0,t) \rangle} \right)} = \pm \sigma_{y} \sqrt{2 \ln \left( \frac{\langle C(x,0,0,t) \rangle}{\langle C^{*} \rangle} \right)}$$

### 13. Plume, Source at H<sub>r</sub>, *u* Constant



### **Model Implementation**

- **4** Plume *C<sub>max</sub>*: release position
- Puff C<sub>max</sub>: center of cloud
- If atmosphere conditions not known, assume worst case for highest C.
- If wind speed not known, assume 2 m/s
- Consider P-G model assumptions: neutral buoyancy, turbulent mixing, time concentrations (10 min), 0.1 - 10 km distances

### **Britter-McQuaid Dense Gas Model**

- **Ground level releases; rural, flat terrain**
- Atmospheric stability effects not included
- Mixing from drop by gravity of effluent into air
- Main parameters: initial buoyancy, g<sub>o</sub>, initial volume flux, q<sub>o</sub>, or total initial volume, V<sub>o</sub>, wind speed at 10 m elevation, u

$$g_o = g(\rho_o - \rho_a) / \rho_a$$

#### $\rho_a$ = density of ambient air

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### **Applicability of B-M Model**

Plume	Puff Information	
$\left(\frac{g_o q_o}{u^3 D_c}\right)^{1/3} \ge 0.15$	$\frac{\sqrt{g_o V_o}}{u D_i} \ge 0.20$	buoyancy∗amnt/ <i>u</i>
$D_c = \left(\frac{q_o}{u}\right)^{1/2}$	$D_i = V_o^{1/3}$	source dimension
$\frac{u R_d}{x} \ge 2.5$	$\frac{u R_d}{x} \le 0.6$	<i>R</i> <sub>d</sub> , release duration

If model criteria satisfied, use Figs 5-13, 5-14 or Tabs 5-4, 5-5 to est. C or downwind distance, x

### **Implementation of B-M Model**

**↓**  $C_o = 1$  for pure material initially released **↓**  $C_m / C_o$  : ratio of material conc in air to pure

$$q_o = q_L \frac{\rho_L}{\rho_V}$$
  $q_L$ : liquid volumetric discharge rate

 $V_o = q_o R_d$ : initial volume, Puff

Adjust conc for density at  $T_a$ :



 $C_e$  : effective conc  $C^*$  : unadjusted conc  $T_o$  : T at release, K





**Table 5-4**Equations Used to Approximate the Curves in theBritter-McQuaid Correlations Provided in Figure 5-13 for Plumes

Concentration ratio ( <i>C</i> m/ <i>C</i> o)	Valid range for $\alpha = \log \left( \frac{g_0^2 q_0}{u^5} \right)^{1/5}$	$\beta = \log \left[ \frac{x}{\left( q_{\rm o}/u \right)^{1/2}} \right]$
0.1	$\alpha \le -0.55$ -0.55 < $\alpha \le -0.14$ -0.14 < $\alpha \le 1$	1.75 $0.24\alpha + 1.88$ $0.50\alpha + 1.78$
0.05	$\alpha \le -0.68$ -0.68 < $\alpha \le -0.29$ -0.29 < $\alpha \le -0.18$ -0.18 < $\alpha \le 1$	1.92 $0.36\alpha + 2.16$ 2.06 $-0.56\alpha + 1.96$
0.02	$\alpha \le -0.69$ $-0.69 < \alpha \le -0.31$ $-0.31 < \alpha \le -0.16$ $-0.16 < \alpha \le 1$	2.08 $0.45\alpha + 2.39$ 2.25 $-0.54\alpha + 2.16$

Table 5-4Equations Used to Approximate the Curves in theBritter-McQuaid Correlations Provided in Figure 5-13 for Plumes

Concentration ratio ( <i>C</i> m/ <i>C</i> o)	Valid range for $ \alpha = \log \left( \frac{g_o^2 q_o}{u^5} \right)^{1/5} $	$\beta = \log \left[ \frac{x}{(q_{\rm o}/u)^{1/2}} \right]$
0.01	$\alpha \le -0.70$ -0.70 < $\alpha \le -0.29$ -0.29 < $\alpha \le -0.20$ -0.20 < $\alpha \le 1$	2.25 $0.49\alpha + 2.59$ 2.45 $-0.52\alpha + 2.35$
0.005	$\alpha \le -0.67$ -0.67 < $\alpha \le -0.28$ -0.28 < $\alpha \le -0.15$ -0.15 < $\alpha \le 1$	2.40 $0.59\alpha + 2.80$ 2.63 $-0.49\alpha + 2.56$
0.002	$\alpha \leq -0.69$	2.6
0.002	$-0.69 < \alpha \leq -0.25$	$0.39\alpha + 2.87$
0.002	$-0.25 < \alpha \leq -0.13$	2.77
0.002	$-0.13 < \alpha \le 1$	$-0.50\alpha + 2.71$

Concentration ratio (C <sub>m</sub> /C <sub>o</sub> )	Valid range for $\alpha = \log \left(\frac{g_o V_o^{1/3}}{u^2}\right)^{1/2}$	$\beta = \log\left(\frac{x}{V_{o}^{1/3}}\right)$
0.1	$\alpha \le -0.44$ -0.44 < $\alpha \le 0.43$ $0.43 < \alpha \le 1$	0.70 $0.26\alpha + 0.81$ 0.93
0.05	$\alpha \le -0.56$ -0.56 < $\alpha \le 0.31$ $0.31 < \alpha \le 1.0$	0.85 $0.26\alpha + 1.0$ $-0.12\alpha + 1.12$
0.02	$\alpha \le -0.66$ -0.66 < $\alpha \le 0.32$ $0.32 < \alpha \le 1$	0.95 $0.36\alpha + 1.19$ $-0.26\alpha + 1.38$
0.01	$\alpha \le -0.71$ -0.71 < $\alpha \le 0.37$ $0.37 < \alpha \le 1$	1.15 $0.34\alpha + 1.39$ $-0.38\alpha + 1.66$
0.005	$\alpha \le -0.52$ -0.52 < $\alpha \le 0.24$ $0.24 < \alpha \le 1$	1.48 $0.26\alpha + 1.62$ $0.30\alpha + 1.75$
0.002	$\alpha \le 0.27$ $0.27 \le \alpha \le 1$	$1.83 - 0.32\alpha + 1.92$
0.001	$\alpha \le -0.10$ $-0.10 < \alpha \le 1$	$2.075 -0.27\alpha + 2.05$

**Table 5-5**Equations Used to Approximate the Curves in theBritter-McQuaid Correlations Provided in Figure 5-14 for Puffs

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## **Toxic Effect Criteria**

- A Normal work hours criteria: TLV-TWA (ACGIH), PEL (OSHA)
- Probit correlations for wide ranges of concentrations and exposure times
- Criteria for short term exposures at higher than TLV-TWA values: available from many sources
- IDLH (NIOSH), 30 min exposures: SCBA required for higher levels

### **ERPG Toxic Effect Criteria**

- American Industrial Hygiene Association (AIHA): Emergency response planning guidelines (ERPG) for exposures up to 1 hour
- ERPG-1: mild transient effects
- ERPG-2: reversible health effects
- **4** ERPG-3: without life-threatening effects
- **4** Tab 5-6, pp. 201, 202
- Alternative guidelines in lieu of ERPG data: Tab 5-9, p 206
- EPA Toxic Endpoints based on ERPG-2

#### Table 5-6 (continued)

Chemical	ERPG-1	ERPG-2	ERPG-3
Methyl isocyanate	0.025	0.5	5
Methyl mercaptan	0.005	25	100
Methyltrichlorosilane	0.5	3	15
Monomethylamine	10	100	500
Perfluoroisobutylene	NA	0.1	0.3
Phenol	10	50	200
Phosgene	NA	0.2	1
Phosphorus pentoxide	5 mg/m <sup>3</sup>	25 mg/m <sup>3</sup>	100 mg/m <sup>3</sup>
Propylene oxide	50	250	750
Styrene	50	250	1,000
Sulfonic acid (oleum, sulfur			
trioxide, and sulfuric acid)	$2 \text{ mg/m}^3$	$10 \text{ mg/m}^3$	30 mg/m <sup>3</sup>
Sulfur dioxide	0.3	3	15
Tetrafluoroethylene	200	1000	10,000
Titanium tetrachloride	5 mg/m <sup>3</sup>	$20 \text{ mg/m}^3$	100 mg/m <sup>3</sup>
Toluene	50	300	1000
Trimethylamine	0.1	100	500
Uranium hexafluoride	5 mg/m <sup>3</sup>	15 mg/m <sup>3</sup>	30 mg/m <sup>3</sup>
Vinyl acetate	5	75	500

Primary guideline	Hierarchy of alternative guidelines	Source	
ERPG-3		AIHA	
	EEGL (30-minute)	NRC	
	IDLH	NIOSH	
ERPG-2		AIHA	
	EEGL (60 minute)	NRC	
	LOC	EPA/FEMA/DOT	
	PEL-C	OSHA	
	TLV-C	ACGIH	
	$5 \times TLV$ -TWA	ACGIH	
ERPG-3		AIHA	
	PEL-STEL	OSHA	
	TLV-STEL	ACGIH	
	$3 \times TLV$ -TWA	ACGIH	

#### Table 5-9 Recommended Hierarchy of Alternative Concentration Guidelines<sup>1</sup>

### **EEGL Toxic Effect Criteria**

- A National Research Council (NRC): Emergency exposure guidance levels (EEGL)
- Acceptable exposure levels for emergency condition tasks up to 1 or up to 24 hours
- Includes reversible effects that do not impair work performance
- **4** Tab 5-7, p 204

**Table 5-7**Emergency Exposure Guidance Levels (EEGLs) from the NationalResearch Council (NRC) (all values are in ppm unless otherwise noted)

Compound	1-hr EEGL	24-hr EEGL	Source
Acetone	8500	1000	NRCI
Acrolein	0.05	0.01	NRCI
Aluminum oxide	15 mg/m <sup>3</sup>	100	NRC IV
Ammonia	100		NRC VII
Arsine	1	0.1	NRC I
Benzene	50	2	NRC VI
Bromotrifluoromethane	25,000		NRC III
Carbon disulfide	50		NRC I
Carbon monoxide	400	50	NRC IV
Chlorine	3	0.5	NRC II
Chlorine trifluoride	1		NRC II
Chloroform	100	30	NRC I
Dichlorodifluoromethane	10,000	1000	NRC II
Dichlorofluoromethane	100	3	NRC II
Dichlorotetrafluoroethane	10,000	1000	NRC II
1,1-Dimethylhydrazine	0.241	0.01 1	NRC V

## **Release Mitigation**

- Part of consequence modeling, Fig 4-1, p 110. Mitigation methods, Tab 5-10, p 214
- Mitigation measures depend on likelihood of a release
- Preventive: Inherent safety, process and mechanical integrity, training, maintenance, sensors, software
- Protective, reduce effect of incidents: curtains, foams, emergency response program

Major area	Examples
Inherent safety	Inventory reduction: Less chemicals inventoried or less in process vessels Chemical substitution: Substitute a less hazardous chemical for one more hazardous Process attentuation: Use lower temperatures and pressures
Engineering design	Plant physical integrity: Use better seals or materials of construction Process integrity: Ensure proper operating conditions and material purity Process design features for emergency control: Emergency relief systems Spill containment: Dikes and spill vessels
Management	Operating policies and procedures Training for vapor release prevention and control Audits and inspections Equipment testing Maintenance program Management of modifications and changes to prevent new hazards Security

#### Table 5-10 Release Mitigation Approaches<sup>1</sup>

Major area	Examples
Early vapor detection and warning	Detection by sensors Detection by personnel
Countermeasures	Water sprays Water curtains Steam curtains Air curtains Deliberate ignition of explosive cloud Dilution Foams
Emergency response	On-site communications Emergency shutdown equipment and procedures Site evacuation Safe havens Personal protective equipment Medical treatment On-site emergency plans, procedures, training, and drills

#### Table 5-10 Release Mitigation Approaches<sup>1</sup>

<sup>1</sup>Richard W. Prugh and Robert W. Johnson, *Guidelines for Vapor Release Mitigation* (New York: American Institute of Chemical Engineers, 1988).