

Chapter 1. Introduction

Polymer(many parts)

: long chain molecule composed of a large number of repeating units

polymer (n>1000) : plastics, fibers and elastomers

oligomer (n<10)

U.S. polymer production (1993)

: plastics(78%), fibers(14%), elastomers(7.4%)

1.1 Classification

Thermal processing behavior : thermoplastics & thermosets

Polymerization mechanism : addition & condensation

ex) addition-type polymers derived from substituted ethylene

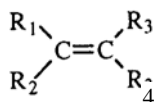



TABLE 1.2 EXAMPLES OF SOME IMPORTANT ADDITION POLYMERS DERIVED FROM ETHYLENE DERIVATIVES

Polymer	R ₁	R ₂	R ₃	R ₄	Repeating Unit
Polyethylene	H	H	H	H	$[-CH_2-CH_2-]$
Polypropylene	H	H	H	CH ₃	$[-CH_2-\underset{\text{CH}_3}{\text{CH}}-]$
Poly(vinyl chloride)	H	H	H	Cl	$[-CH_2-\underset{\text{Cl}}{\text{CH}}-]$
Poly(vinyl alcohol)	H	H	H	OH	$[-CH_2-\underset{\text{OH}}{\text{CH}}-]$
Polyacrylonitrile	H	H	H	C≡N	$[-CH_2-\underset{\text{C}\equiv\text{N}}{\text{CH}}-]$
Poly(vinyl acetate)	H	H	H	$\begin{array}{c} \\ \text{O} \\ \\ \text{C}=\text{O} \\ \\ \text{CH}_3 \end{array}$	$[-CH_2-\underset{\begin{array}{c} \\ \text{O} \\ \\ \text{C}=\text{O} \\ \\ \text{CH}_3 \end{array}}{\text{CH}}-]$
Polystyrene	H	H	H		$[-CH_2-\underset{\text{C}_6\text{H}_5}{\text{CH}}-]$
Poly(methyl methacrylate)	H	H	CH ₃	$\begin{array}{c} \\ \text{C}=\text{O} \\ \\ \text{O} \\ \\ \text{CH}_3 \end{array}$	$[-CH_2-\underset{\begin{array}{c} \text{CH}_3 \\ \\ \text{C} \\ \\ \text{C}=\text{O} \\ \\ \text{O} \\ \\ \text{CH}_3 \end{array}}{\text{C}}-]$
Poly(vinylidene chloride)	H	H	Cl	Cl	$[-CH_2-\underset{\text{Cl}}{\overset{\text{Cl}}{\text{C}}}-]$

Tacticity : isotactic, syndiotactic & atactic

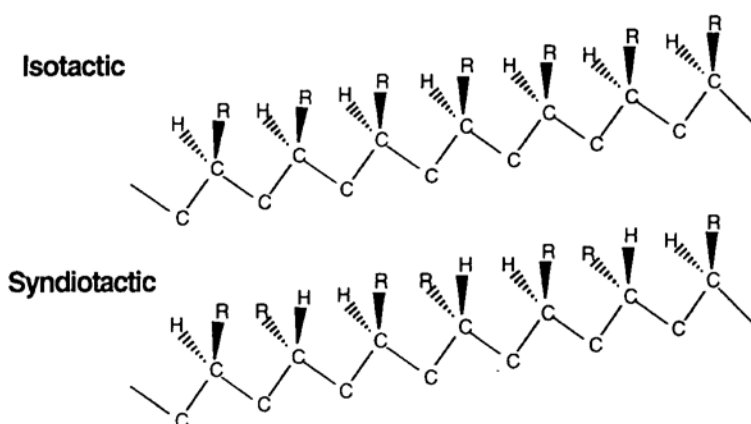


Figure 1.5. Two forms of stereochemical configuration of an extended-chain vinyl polymer having a substituent group R other than hydrogen.

Geometric isomers

ex) polybutadiene : vinyl, cis & trans

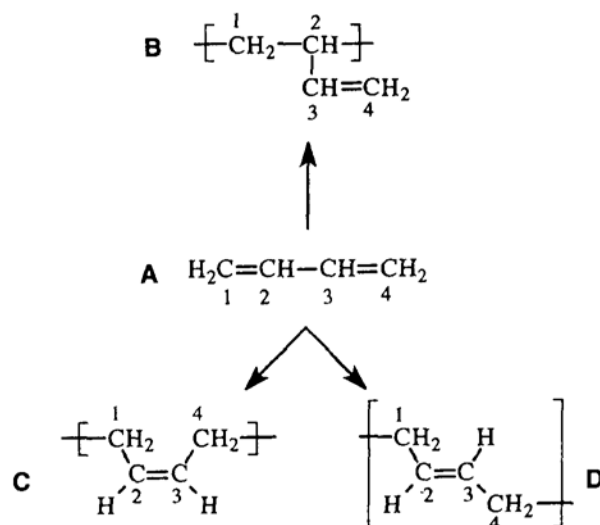


Figure 1.7. Alternative pathways for the polymerization of 1,3-butadiene (A) to give 1,2-poly(1,3-butadiene) (B), *cis*-1,4-poly(1,3-butadiene) (C), or *trans*-1,4-poly(1,3-butadiene) (D).

Nomenclature

structure-based : poly(CRU) --- IUPAC name

source-based : poly(monomer) --- popular

ex) polystyrene --> poly(1-phenylethylene)

polytetrafluoroethylene --> poly(difluoromethylene)

1.3 Molecular Weight

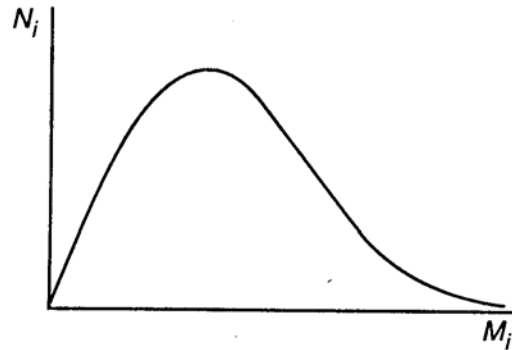


Figure 1.8. A typical distribution of molecular weights shown as a plot of the number of moles of chains, N_i , having molecular weight, M_i , against M_i .

Average MW

$$\bar{M} = \frac{\sum N_i M_i^\alpha}{\sum N_i M_i^{\alpha-1}} \quad (\text{discrete distribution}) \quad W_i = N_i M_i$$

$\alpha = 1$: number avg. MW

$\alpha = 2$: weight avg. MW

$\alpha = 3$: z avg. MW

$$\text{--->} \quad \bar{M} = \frac{\int_0^\infty N M^\alpha dM}{\int_0^\infty N M^{\alpha-1} dM} \quad (\text{continuous distribution})$$

$$\text{PDI (polydispersity index)} = \frac{\bar{M}_w}{\bar{M}_n}$$

1.4 Thermal Transition

$$150\text{K} < T_g < 770\text{K} \quad 334\text{K} < T_m < 675\text{K}$$

In general, $T_g \approx 1/2 - 2/3$ of T_m

Example problem 1-1.

A polydisperse sample of polystyrene is prepared by mixing three monodisperse samples in the following proportions:

1g	10,000 molecular weight
2g	50,000 molecular weight
2g	100,000 molecular weight

Using this information, determine the following: (a) number-average; (b) weight-average; and (c) z-average molecular weight of the mixture.