Investigation of NO reduction by CO reaction over NO_x/CeO_2 catalysts

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 CeO_2 supported NO_x catalysts (NO_x/CeO₂) have been widely studied in various catalytic reactions including NO reduction by CO. In this work, a series of NO_x/CeO₂ catalysts were prepared and tested with various characterizations techniques (e.g., BET, Raman, XRD, DRIFT) as well as Gas Chromatography (GC) to investigate their physicochemical properties, surface properties, and catalytic activities in the NO reduction by CO reaction. In addition to the nickel loading, the impact of catalyst synthesis conditions on the molecular/electronic structure and the catalytic performance were studied. The increase in N loading of the catalyst (up to monolayer) led to decrease in specific surface area, formation of NO_x crystalline structures on CeO₂ surface, easier reduction of the catalyst comparing to bulk NO_x and bulk CeO₂, as well as increase in catalytic activity. It was also concluded

that the presence of oxygen vacancy/defect site, N^{2+} oxidation state, and smaller crystallite size are believed to enhance the catalytic activity. The results provided insights on the structure-activity relationship of NO_x/CeO_2 catalysts for NO reduction by CO reaction.