

Mechanical study of CO₂ sorption on salt-promoted MgO-CeO₂ under practical gas conditions
in the presence of H₂O

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The enhancement mechanism of H₂O dissociation-promoted CO₂ sorption on the MgCeO_x surface was studied through in situ DRIFTS analysis. Among solid CO₂ sorbents, MgO attracts attention due to its applicability at high pressure and intermediate temperature. Notably, it has been found that H₂O promotes sorption performance but still lacks insight into the dynamic sorption process under the corresponding conditions. Herein, the main contributor and mechanism to the CO₂ sorption in each condition were analyzed through the in situ DRIFTS results measured with the injection of a mixture gas of CO₂, H₂O, and Ar. The adsorbed OH groups grew significantly with monodentate carbonate at the early sorption stage, leading to H₂O dissociation-promoted sorption process. This study reveals the contribution of H₂O to the performance improvement of the sorbent when applying the MgO-based CO₂ sorbent to the actual process.