

Comparison of the performance of chemical looping combustion process using Fe-Mn low-cost particles

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In Chemical Looping combustion, a redox reaction occurs in two separate reactors. As a result, CO<sub>2</sub> can be separated from the process without any additional CO<sub>2</sub> capture facility, and NO<sub>x</sub> generated during the high temperature combustion process is blocked, and combustion efficiency is also high. In this study, Fe-Mn-based particles using Al<sub>2</sub>O<sub>3</sub> as a support were used for the experiment. In the case of Fe, Fe<sup>2+</sup> is not effective in the reaction with methane, but Fe<sup>3+</sup> is very effective for oxygen transfer. Therefore, the precursor was composed of Me<sup>3+</sup> containing oxides, and the spinel structure, which is structurally stable Me<sup>3+</sup>, was formed and used. Particle efficiency was analyzed by TGA and XPS was used to analyze the oxidation state. XRD, SEM/EDS, and CV were used for analysis of other physicochemical properties. Thus, it was confirmed that sufficient oxygen transfer ability was obtained even if low cost particles were used.