

Facile synthesis of dendrite-structured iridium oxide as electrocatalyst for oxygen evolution reaction

김수찬, 이상하, 조미숙, 이영관<sup>†</sup>

성균관대학교

(ykleee@skku.edu<sup>†</sup>)

Iridium oxide ( $\text{IrO}_x$ ) is one of the best electrocatalysts for the oxygen evolution reaction in acidic environments. The activity of the electrocatalyst depends on the nature of material, surface area, morphology and so on. To enhance electrocatalytic activity of the electrode material,  $\text{IrO}_x$  is electrochemically deposited on gold dendrite, which is synthesized by chronoamperometry. Gold dendrite provides  $\text{IrO}_x$  electrode with high surface area and good conductivity. As prepared,  $\text{IrO}_x$  dendrite is well-defined and has high surface area. The morphology and composition of  $\text{IrO}_x$  are investigated by scanning electron microscope, X-ray diffraction analysis, and energy dispersive spectrometer. Electrochemical properties of the electrode is characterized by cyclic voltammetry and linear sweep voltammetry. The prepared electrode shows a high electrocatalytic activity towards the oxygen evolution with low overpotential.