The direct deposition of thermally treated graphene oxide/TiO<sub>2</sub> porous material on the coin cell anode for lithium-ion battery anode

<u>Balasubramaniyan</u>, 김병수<sup>1</sup>, 손혜정<sup>1</sup>, 정진석<sup>1,†</sup> 울산대학교; <sup>1</sup>울산대학교 화학공학부 (jschung@mail.ulsan.ac.kr<sup>†</sup>)

The anatase–TiO<sub>2</sub>, it has been identified as the most promising anode material for lithium–ion batteries (LIB) due to its negligible volume change, and no solid electrolyte interface formation. Recently, the drawbacks of poor electronic conductivity and lower electrolyte diffusion of the TiO<sub>2</sub> has been resolved by doping, or hybrid with carbon based materials. Herein, the thermally treated graphene oxide/TiO<sub>2</sub> (TGO/TiO<sub>2</sub>) is prepared through the calcination of the poly (methylmethacrylate)–GO/TiO<sub>2</sub> (PMMA–GO/TiO<sub>2</sub>) on the coin cell anode. The decomposition of PMMA increased the surface area of pristine TiO<sub>2</sub> of 10.3 m<sup>2</sup> g<sup>-1</sup> to 182.3 m<sup>2</sup> g<sup>-1</sup> of TGO/TiO<sub>2</sub>. In LIB anode, the TGO/TiO<sub>2</sub> porous electrode exhibited much higher specific capacity of 354 mAh g<sup>-1</sup> at 0.2C than pristine TiO<sub>2</sub> of 159.2 mAh g<sup>-1</sup> and showed an excellent rate capabilities at all the C-rates. The higher specific capacity of TGO/TiO<sub>2</sub> electrode can be attributed that the highly porous characteristics and their resulting pseudocapacitive effect of the nanocomposite electrode during the Li–ion insertion/extraction reactions.