

Multiple hydrogen molecular loading into sII hydrate induced by water-soluble sH former

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Hydrogen storage in clathrate hydrate imposes various important physical processes occurring in the inclusion compounds. Multiple H₂ occupation in the nano-sized cage formed by hydrogen-bonded water framework shows the possibility of clathrate hydrate as future hydrogen storage materials. We present here, for the first time, H₂ molecular loading into structure II (sII) hydrate induced by water-soluble sH former (1-methylpiperidine) as evidenced by PXRD and Raman spectroscopic measurements. Structural change is accompanied by reducing the concentration of 1-methylpiperidine, as the concentration decreased below 2.8 mol% and structure transition from structure H to the favorable structure II proceeds. In the course of structural transition, hydrate structure is fully tuned to adopt hydrogen clusters (2-4 H₂ molecules) in 51264 cages and single occupancy in 512 cages which significantly increase hydrogen storage wt% by 40-60% compared to sH hydrate.