

Low-temperature vacuum stripping of CO<sub>2</sub> from aqueous amine solutions using thin-film silicalite-filled PDMS composite membranes

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Carbon dioxide (CO<sub>2</sub>) recovery by vacuum stripping technology using membranes was studied as an alternative to the desorption process at low temperatures (below 120°C) in conventional aqueous amine absorption process. Composite membranes were prepared by coating hydrophobic silicalite-filled polydimethylsiloxane (PDMS) layers on polyethylene (PE) porous supports and used as new membrane strippers for CO<sub>2</sub> recovery to prevent the typical pore wetting problem of hydrophobic porous membranes. Aqueous 30 wt% solutions of monoethanolamine (MEA), diethanolamine (DEA). CO<sub>2</sub> stripping fluxes increased with increasing temperature and CO<sub>2</sub> loading of amine solutions as well as with decreasing stripping pressures because of the enhanced upstream and downstream driving forces. CO<sub>2</sub> stripping fluxes were much higher in the TEA and DEA solutions than in the MEA solutions. The thin-film silicalite-filled PDMS composite membranes showed excellent long-term stability in the vacuum stripping process when compared with porous polytetrafluoroethylene (PTFE) membranes.