

Isobaric Vapor-Liquid equilibria for the Ethylene glycol monopropyl ether (C3E1) + 2-Propanol  
and  
2-Butanol systems

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This paper shows isobaric vapor-liquid equilibria (VLE) for the binary systems of Ethylene glycol monopropyl ether (C3E1) + 2-Propanol and 2-Butanol. Alkoxyethanol such as C3E1 has associating groups so that it shows a highly non-ideal phase behavior due to the H-bond interaction. The experiments of this isobaric VLE system were carried out at different pressures for the measurement of the associating fluid mixture by Fischer VLE 602. In this system, we investigated the correlation for the experimented data using two term virial equation for vapor-phase fugacity coefficients and the three suffix Margules equation, Wilson, NRTL, UNIQUAC for liquid-phase activity coefficients. The results of the experiment shows a good agreement with variety of models.