A Kinetic Study of the Oxidation of UO₂ to U₃O₈

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In this work, a kinetic model has been suggested to explain the mass transfer and the diffusion effects during the oxidation of a UO2 pellet. The results calculated by the model equation are in good agreement with the experimental data. The kinetic parameters were estimated by a fitting of the experimental data. The resistances from the mass transfer and the pore diffusion increase with an increase of the fractional oxidation. The oxidation characteristics of UO2 were interpreted by the suggested kinetic model.