

Structural Transition and Tuning of the *tert*-Butylamine Hydrate for Enhanced Methane Storage Capacity

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Although many researches are performing vividly for utilizing clathrate hydrates in the area of the gas transportation and storage, they are mainly focused on using water molecules as a lattice structure. In this study, we used *t*-BuNH₂ as host molecules to accommodate CH₄ guest molecules. The structural transition and the coexistence of structures according to the initial concentration of *t*-BuNH₂ are identified by means of powder X-ray diffraction (PXRD) and nuclear magnetic resonance (NMR), and the distributions of CH₄ molecules in cages of the double hydrates are also analyzed. These results lead us to think that the storage capacity of guest molecules can be tuned by using the structural transition. These microscopic results and the proposed tuning mechanism reported here can shed light on the future applications of the double hydrate in the area of the natural gas storage and transportation.