Gasification of wood using a thermogravimetric analyzer

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Gasification of wood including larch, oak, and apple was conducted using a thermogravimetric analyzer. A major reaction variable was final temperature ranging from 600 to 900 $^{\circ}$ C. Reaction temperatures were enhanced to the final temperatures at the temperature increasing rate of 20 $^{\circ}$ C/min under a nitrogen flow environment, and then held at the isothermal conditions for 30 min, followed by isothermal combustions at the air flow rate of 90 ml/min. Thermolysis of the reactants was initiated at temperatures around 100 $^{\circ}$ C. The temperature range at which the highest reactant reduction rate appeared was different with the feedstocks. A kinetic study of the experimental results obtained at the isothermal gasification conditions was carried out to determine reaction rate constant and activation energy.