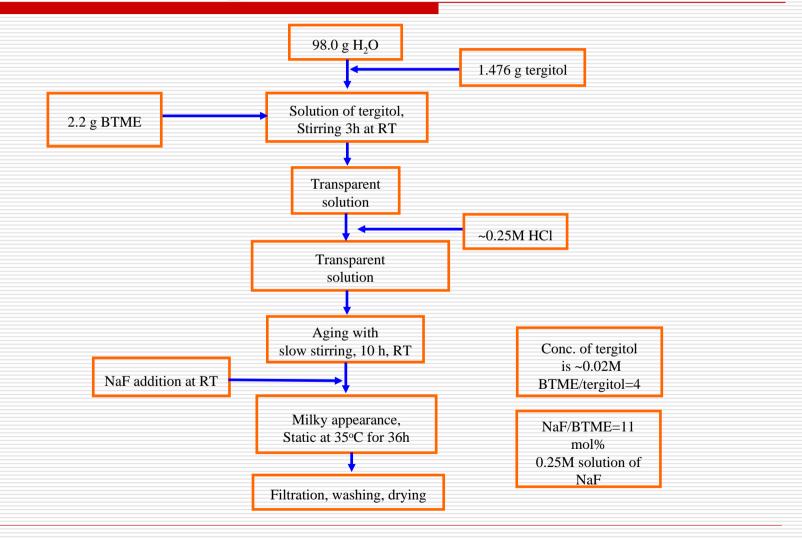
# Synthesis of spherical hybrid mesoporous silica of MSU-1 type using BTME by the two-step process

## Synthesis steps



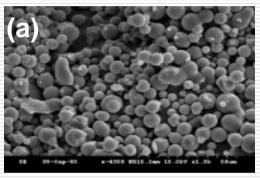
Catalysis and Nano-materials Lab. INHA University

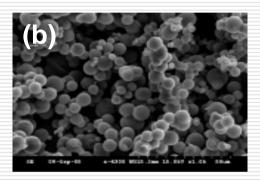
#### Effect of mol% of NaF

(I)

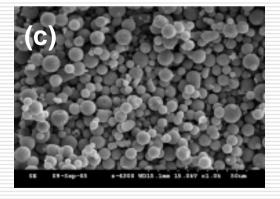
#### **Effect of NaF mole percent over BTME**

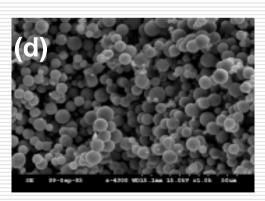
(a) 8 mol%, (b) 9 mol%, (c) 10 mol% and (d) 11 mol%





# Comment 1)Similar morphology for 8-15 mol% of NaF 2)Compare with TEOS (2mol%) as a silica source, substantially larger amount of F<sup>-</sup> is required for BTME.

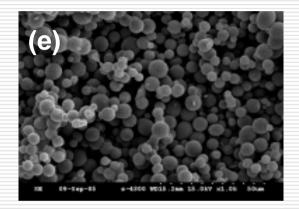


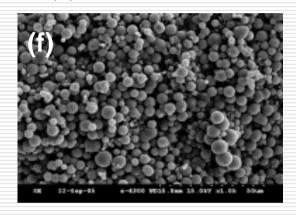


#### Effect of mol% of NaF

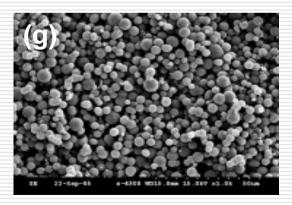
(II)

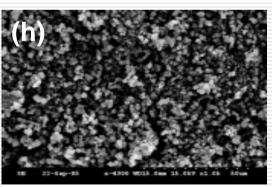
SEM images of the samples prepared with varying mol% of NaF (e) 12 %, (f) 14 %, (g) 15 % and (h) 20 %





For 20 mol% of NaF No good morphology

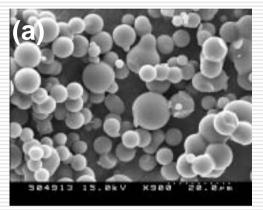




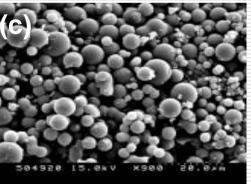
#### Effect of temperature

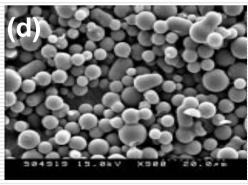
#### **Hydrolysis/Condensation temperature effect**

(a) RT/35 , (b) 35/35 , (c) 35/45 , and (d) 35/55





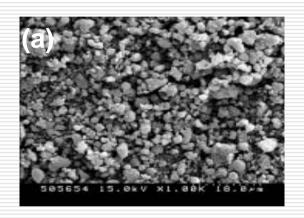


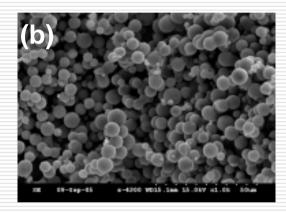


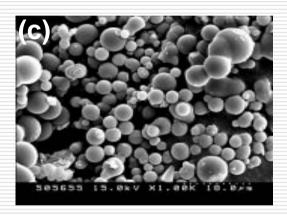
Hydrolysis at RT and condensation at 35 is better for the desired spherical morphology with size distribution in the range of 2 to 8 µm.

#### **Effect of BTME concentration**

Effect of different BTME over surfactant mol ratios (a) 2, (b) 4, and (c) 6



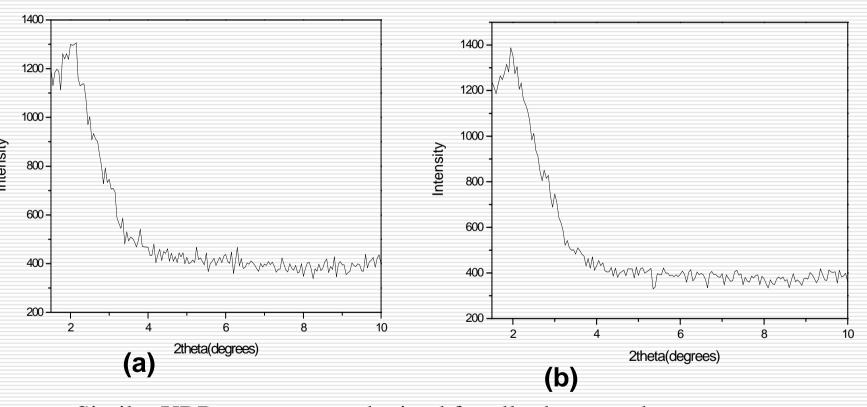




To obtain desired morphology the optimum mol ratio of BTME: surfactant = 4, whereas in the case of TEOS the optimum mol ratio of TEOS: surfactant = 8.

### XRD patterns of the as-made sample

Small angle XRD peaks of (a)10 mol% of NaF, (b) 15 mol % of NaF



Similar XRD patterns are obtained for all other samples

#### Conclusion

To obtain a narrow particle size distribution in the range of 2-8 µm, optimum temperature of hydrolysis and condensation steps are RT and 35 , respectively. Spherical morphology is obtained for 8-15 mol% of NaF. The optimum mol ratio of BTME: surfactant was 4.