CHBE320 LECTURE VII DYNAMIC BEHAVIORS OF REPRESENTATIVE PROCESSES (II)

Professor Dae Ryook Yang

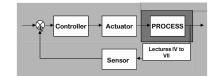
Fall 2021 Dept. of Chemical and Biological Engineering Korea University

CHBE320 Process Dynamics and Control

Korea University 7-1

Road Map of the Lecture VII

- Dynamic Behavior of Representative Processes
 - Inverse response
 - Time delays
 - High-order processes
 - Interacting-Noninteracting processes
 - Distributed parameter system
 - MIMO process



CHBE320 Process Dynamics and Control

Korea University 7-2

INVERSE RESPONSE IN CHEMICAL PROCESSES

- · Reboiler level to the change in boilup rate
 - If boilup rate is increase, more liquid will vaporized and the inventory in the reboiler will be reduced and the level will be decreased.
 - However, at the beginning of the boilup rate increase, more vapor is generated and the vapor will flow upward to the stage above. In that stage, more vapor will pass through the liquid and the density of liquid will decrease so that the more liquid will spill over the weir. This results a temporary increase in the level of reboiler.
 - Eventually, the density of liquid settles down and overflow will reach at another steady state. Then the reboiler level will be decreased.

CHBE320 Process Dynamics and Control

Korea University 7-3

Bottom

Product

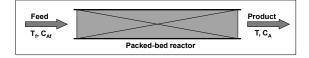
Seal pan

Steam

Condensate

Outlet Temp of the exothermic packed bed reactor to feed temp change

- An increase in feed temp will speed up the reaction rate and the reactor outlet temp will increase due to the increased reaction heat generated.
- However, at the beginning of the feed temp increase, more reaction occurs in the inlet part of the reactor and more reactants are consumed. This causes a decrease of reactant concentration in the outlet part of the reactor, and the outlet temp will decrease due to temporary drop in reactant concentration.
- Eventually, increase in feed temp will enhance the reaction and generate more reaction heat and the outlet temp will increase slowly.

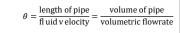


CHBE320 Process Dynamics and Control

Korea University 7-4

TIME DELAYS

- Fluid transportation through a pipe
 - Also, called distance-velocity lag, transportation lag, dead time

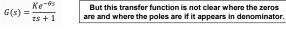


$\overrightarrow{}$ v		* *
•	L	

Transfer function

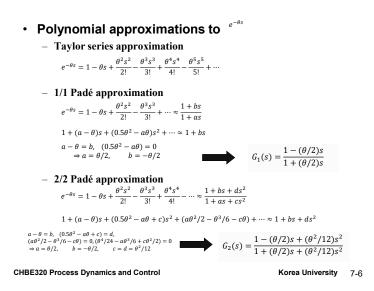
 $y(t) = \begin{cases} 0 & \text{for } t < \theta \\ x(t-\theta) & \text{for } t \ge \theta \end{cases} \implies \frac{Y(s)}{X(s)} = G(s) = e^{-\theta s}$

· Many high-order systems can be approximated by a first-order plus dead-time model (FOPDT).



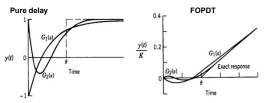
CHBE320 Process Dynamics and Control

Korea University 7-5



Taylor series approximation

- Increase the order of numerator (physical realizability)
- AR and phase angle are different from the exact
- Low accuracy
- Padé approximation
 - Does not change the order of transfer function
 - Only phase angle is different
 - Higher accuracy
 - Oscillatory behavior (complex poles and zeros)



CHBE320 Process Dynamics and Control

Korea University 7-7

APPROXIMATION OF HIGHER-ORDER SYSTEMS

Approximation by FOPDT or SOPDT models

$$G(s) = \frac{K}{(\tau_1 s + 1)(\tau_2 s + 1)\cdots(\tau_n s + 1)}$$

If $\tau_1 \gg (\tau_2, \dots, \tau_n)$, $(\tau_1$ is the "dominant time constant")

$$G(s) \approx \frac{Ke^{-\theta}}{(\tau_1 s + 1)}$$
 where $\theta = \tau_2 + \dots + \tau_n$

If $\tau_1, \tau_2 \gg (\tau_3, \cdots, \tau_n)$, $(\tau_1$ and τ_2 are the "dominant time constants")

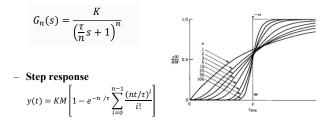
$$G(s) \approx \frac{Ke^{-\theta s}}{(\tau_1 s + 1)(\tau_2 s + 1)}$$
 where $\theta = \tau_3 + \dots + \tau_n$

CHBE320 Process Dynamics and Control

Korea University 7-8

 \circ

nth-order system with equal time constants



By deinition, $e \triangleq \lim_{x \to 0} (1+x)^{1/x} = 2.7182818285...$

$$\lim_{n \to \infty} G_n(s) = \lim_{n \to \infty} \frac{K}{(\frac{t}{n}s+1)^n} = \lim_{n \to \infty} \frac{K}{\left[(\frac{ts}{n}+1)^{n/\tau s}\right]^{\tau s}} = \frac{K}{\left[\lim_{x \to 0} (1+x)^{1/x}\right]^{\tau s}} = Ke^{-\tau s}$$

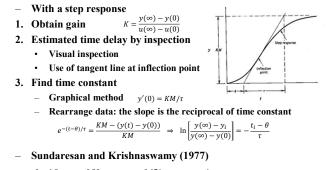
<u>y</u>

CHBE320 Process Dynamics and Control

Korea University 7-9

FITTING DATA TO EMPIRICAL MODELS

• Fitting FOPDT model using step test

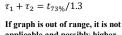


 $\theta = 1.3t_{35.3\%} - 0.29t_{85.3\%} \quad \tau = 0.67(t_{85.3\%} - t_{35.3\%})$

Korea University 7-10



- Harriot's Method for overdamped systems Calculate $y_{t_{73\%}/2.6}$ and read $\tau_1/(\tau_1 + \tau_2)$ from graph

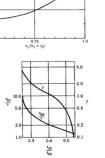


applicable and possibly higherorder or underdamped.

- Smith's method
 - Use t_{20%} and t_{60%}.
 - Read parameters from graph
- Nonlinear/Linear Regression Use of optimization to minimize the error between the data and calculated model value.

$$\min_{K,\tau,\zeta,\theta} \sum_{i=1}^{N} (y(t_i) - y_i)^2 \text{ where } y(t_i) = f(t_i, K, \tau, \zeta, \theta)$$

CHBE320 Process Dynamics and Control



Korea University 7-11

INTERACTING AND NONINTERACTING PROCESSES

Non interacting process

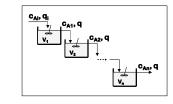
CHBE320 Process Dynamics and Control

- The first tank affects the second tank but second tank does not affect the first tank and so on. This is called "noninteracting".
- Each tank can be modeled as

$$\frac{\mathcal{C}_{Aj}(s)}{\mathcal{C}_{A(j-1)}(s)} = \frac{K_j}{\tau_j s + 1}$$

- For the whole system

$$\frac{C_{An}(s)}{C_{Ai}(s)} = \prod_{j=1}^{n} \frac{K_j}{(\tau_j s + 1)}$$



CHBE320 Process Dynamics and Control

Interacting process

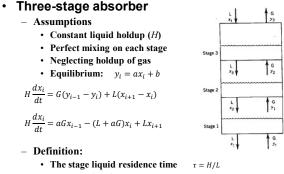
- Many chemical processes exhibit interacting nature. $A_1 \frac{dh_1}{dt} = q_i - q_1 \qquad A_2 \frac{dh_2}{dt} = q_1 - q_2$ $q_1 = \frac{1}{R_1}(h_1 - h_2) \qquad q_2 = \frac{1}{h_2}h_2$ $A_1 \frac{dh_1}{dt} = q_i - \frac{1}{R_1}(h_1 - h_2) \qquad A_2 \frac{dh_2}{dt} = \frac{1}{R_1}(h_1 - h_2) - \frac{1}{R_2}h_2$ $A_1 \frac{dh_1}{dt} = q_i - \frac{1}{R_1}(h_2 - h_2) \qquad A_2 \frac{dh_2}{dt} = \frac{1}{R_1}(h_1 - h_2) - \frac{1}{R_2}h_2$ $A_1 R_1 s \tilde{H}_1(s) + \tilde{H}_1(s) - \tilde{H}_2(s) = R_1 \tilde{Q}_1(s)$ $\frac{A_2 R_1 R_2}{R_1 + R_2} s \tilde{H}_2(s) + \tilde{H}_2(s) = \frac{R_2}{R_1 + R_2} \tilde{H}_1(s) \Rightarrow \frac{\tilde{H}_2(s)}{\tilde{H}_1(s)} = \frac{R_2/(R_1 + R_2)}{A_2 R_1 R_2/(R_1 + R_2) s + 1}$ $\frac{\tilde{H}_2(s)}{\tilde{Q}_1(s)} = \frac{R_2}{A_1 A_2 R_1 R_2 s^2 + (A_1 R_1 + A_2 R_2 + A_1 R_2) s + 1}$ $\frac{\tilde{H}_2(s)}{\tilde{Q}_1(s)} = \frac{R_2}{\tau^2 s^2 + 2\zeta \tau s + 1} \qquad \text{where } \tau = \sqrt{A_1 A_2 R_1 R_2}, \zeta = \frac{(A_1 R_1 + A_2 R_2 + A_1 R_2)}{2\sqrt{A_1 A_2 R_1 R_2}}$

Since arithmetic mean \geq geometric mean, $\zeta > 1$ (overdamped)

CHBE320 Process Dynamics and Control

Korea University 7-13

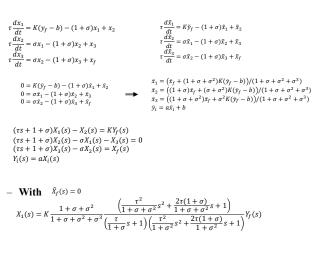
STAGED SYSTEM

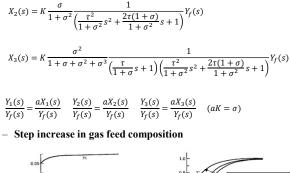


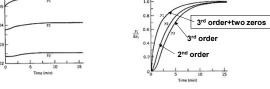
- The stripping factor $\sigma = aG/L$
- The gas-to-liquid ratio K = G/L

CHBE320 Process Dynamics and Control

Korea University 7-14







CHBE320 Process Dynamics and Control

Korea University 7-16

CHBE320 Process Dynamics and Control

· Analysis using transfer function

- Easy to develop
- Types of response can be recognized very conveniently
- Parameter effects can be analyzed through lumped parameter
- Linearization required when the system is nonlinear
 - If the effect of flow rate change, the previous example becomes nonlinear.
 - Due to linearization, new model has to be obtained when the operating condition changes widely.
- For the nonlinear system analysis, numerical integration should be considered
 - Utilize differential equation solvers
 - Non-stiff case: explicit method
 - Stiff case: implicit methods
 - ODE
 - DAE
 - PDE

CHBE320 Process Dynamics and Control

Korea University 7-17

DISTRIBUTED PARAMETER SYSTEMS

Lumped parameter system (ODE)

- Dependent variables depend only on time, not on spatial location
- Perfect mixing assumption eliminates the dependency on
- spatial coordinates and the analysis will be conducted in averaging sense
- All balances are valid around the boundary

Distributed parameter system (PDE)

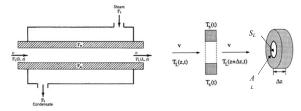
- Dependent variables depend on time and also spatial location
- Perfect mixing is not valid. (double pipe heat exchanger, packed bed reactor

CHBE320 Process Dynamics and Control

Korea University 7-18

DPS EXAMPLE

Double pipe heat exchanger



- Assumptions

- Plug flow (no radial or angular variations in T_L)
- Neglect axial conduction
- · Heat transfer resistance of the inside tube metal is neglected
- · Steam jacket is well mixed

CHBE320 Process Dynamics and Control

Korea University 7-19

• For a control volume, from the energy balance for the liquid

$$\rho_L C_L S_L \Delta z \frac{dT_L(t,z)}{dt} = v \rho_L C_L S_L (T_L(t,z) - T_{ref}) - v \rho_L C_L S_L (T_L(t,z + \Delta z) - T_{ref}) + h_L A_L \Delta z (T_W(t,z) - T_L(t,z))$$

$$\frac{dT_L(t,z)}{dt} = -\frac{v(T_L(t,z+\Delta z) - T_L(t,z))}{\Delta z} + \frac{h_L A_L}{\rho_L C_L S_L} (T_W(t,z) - T_L(t,z))$$

As $\Delta z \rightarrow 0$

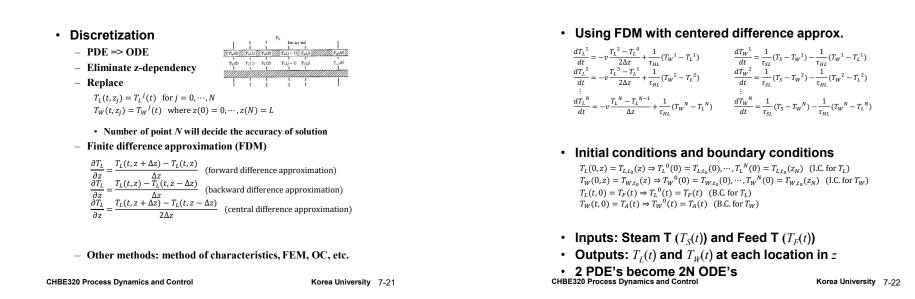
a

$$\frac{dT_L}{dt} = -v \frac{\partial T_L}{\partial z} + \frac{1}{\tau_{HL}} (T_W - T_L) \text{ where } \tau_{HL} = \frac{\rho_L C_L S_L}{h_L A_L}$$

- $(A_L$ is a heat transfer area per unit length)
- For a control volume, from the energy balance for the wall $\frac{dT_W}{dt} = \frac{h_S A_S}{\rho_W C_W S_W} (T_S - T_W) - \frac{h_L A_L}{\rho_L C_L S_L} (T_W - T_L)$

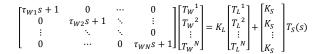
$$\frac{dT_W}{dt} = \frac{1}{\tau_{SW}}(T_S - T_W) - \frac{1}{\tau_{WL}}(T_W - T_L)$$

CHBE320 Process Dynamics and Control



Transfer function

Tranolo	Tuncu	011		Tri-diagonal matrix	
$\begin{bmatrix} \tau_{L1}s + 1 \\ c_1 \\ 0 \\ \vdots \\ 0 \end{bmatrix}$	$b_1 \\ \tau_{L1}s + 1 \\ c_2 \\ \vdots \\ \dots$	$egin{array}{c} 0 \ b_2 \ \ddots \ \ddots \ 0 \end{array}$	c_{N-1}	$\begin{bmatrix} 0 \\ \vdots \\ 0 \\ b_{N-1} \\ \tau_{L1}s + 1 \end{bmatrix} \begin{bmatrix} T_L^1 \\ T_L^2 \\ \vdots \\ T_L^N \end{bmatrix} = K_W \begin{bmatrix} T_W^1 \\ T_W^2 \\ \vdots \\ T_W^N \end{bmatrix} + \begin{bmatrix} K_F \\ 0 \\ \vdots \\ 0 \end{bmatrix} T_F(s)$)



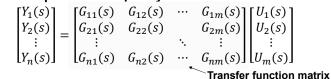
- FDM usually generates tri-diagonal matrix.
- Quite high order system if N is big.
- May not be convenient to analyze.

CHBE320 Process Dynamics and Control

Korea University 7-23

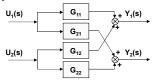
MIMO systems

Multi-Input Multi-Output System



- MISO (n=1) and SIMO (m=1) are possible.





CHBE320 Process Dynamics and Control

- From SS model to TF model for MIMO system
 - With deviation variables after linearization

$$\begin{bmatrix} \dot{x}_1 \\ \dot{x}_2 \end{bmatrix} = \mathbf{A} \begin{bmatrix} x_1 \\ x_2 \end{bmatrix} + \mathbf{B} \begin{bmatrix} u_1 \\ u_2 \end{bmatrix} \quad \Rightarrow \quad \dot{\mathbf{x}} = \mathbf{A}\mathbf{x} + \mathbf{B}\mathbf{u}$$
$$\begin{bmatrix} y_1 \\ y_2 \end{bmatrix} = \mathbf{C} \begin{bmatrix} x_1 \\ x_2 \end{bmatrix} + \mathbf{D} \begin{bmatrix} u_1 \\ u_2 \end{bmatrix} \quad \Rightarrow \quad \mathbf{y} = \mathbf{C}\mathbf{x} + \mathbf{D}\mathbf{u}$$

$$\begin{split} s\mathbf{X}(s) &= \mathbf{A}\mathbf{X}(s) + \mathbf{B}\mathbf{U}(s) \implies (s\mathbf{I} - \mathbf{A})\mathbf{X}(s) = \mathbf{B}\mathbf{U}(s) \\ \mathbf{Y}(s) &= \mathbf{C}\mathbf{X}(s) + \mathbf{D}\mathbf{U}(s) \implies \mathbf{Y}(s) = [\mathbf{C}(s\mathbf{I} - \mathbf{A})^{-1}\mathbf{B} + \mathbf{D}]\mathbf{U}(s) \end{split}$$

$$\therefore \mathbf{G}(s) = [\mathbf{C}(s\mathbf{I} - \mathbf{A})^{-1}\mathbf{B} + \mathbf{D}]$$

- Generally for (pxm) system, A is (nxn), B is (nxm), C is (pxn) and D is (pxm) matrices, respectively.
- Then, G(s) is (pxm) transfer function matrix.

CHBE320 Process Dynamics and Control