# CHBE320 LECTURE IV MATHEMATICAL MODELING OF CHEMICAL PROCESS

**Professor Dae Ryook Yang** 

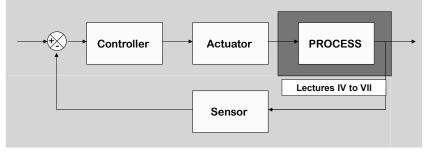
# Fall 2021 Dept. of Chemical and Biological Engineering Korea University

CHBE320 Process Dynamics and Control

Korea University 4-1

# Road Map of the Lecture IV

• Basics of Process Modeling



- Mathematical Modeling
- Steady-state model vs. Dynamic model
- Degree of freedom analysis
- Models of representative processes, etc.

CHBE320 Process Dynamics and Control

# THE RATIONALE FOR MATHEMATICAL MODELING

### · Where to use

- To improve understanding of the process
- To train plant operating personnel
- To design the control strategy for a new process
- To select the controller setting
- To design the control law
- To optimize process operating conditions

### A Classification of Models

- Theoretical models (based on physicochemical law)
- Empirical models (based on process data analysis)
- Semi-empirical models (combined approach)

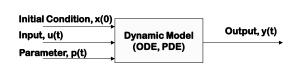
CHBE320 Process Dynamics and Control

Korea University 4-3

# DYNAMIC VERSUS STEADY-STATE MODEL

### Dynamic model

- Describes time behavior of a process
  - Changes in input, disturbance, parameters, initial condition, etc.
- Described by a set of differential equations
  : ordinary (ODE), partial (PDE), differential-algebraic(DAE)



### Steady-state model

- Steady state: No further changes in all variables
- No dependency in time: No transient behavior
- Can be obtained by setting the time derivative term zero

CHBE320 Process Dynamics and Control

## **MODELING PRINCIPLES**

### Conservation law

- Within a defined system boundary (control volume)

[ rate of ]	_	[rate of		rat	te of]		
accumulation	=	[ input ]	_	lou	tput		
+ [		rate of				ate of	1
		generati	disappreanc				

- Mass balance (overall, components)
- Energy balance
- Momentum or force balance
- Algebraic equations: relationships between variables and parameters

CHBE320 Process Dynamics and Control

Korea University 4-5

## **MODELING APPROACHES**

- Theoretical Model
  - Follow conservation laws
  - Based on physicochemical laws
  - Variables and parameters have physical meaning
  - Difficult to develop
  - Can become quite complex
  - Extrapolation is valid unless the physicochemical laws are invalid
  - Used for optimization and rigorous prediction of the process behavior

#### Empirical model

- Based on the operation data
- Parameters may not have physical meaning
- Easy to develop
- Usually quite simple
- Requires well designed experimental data
- The behavior is correct only around the experimental condition
- Extrapolation is usually invalid
- Used for control design and simplified prediction model

CHBE320 Process Dynamics and Control

### **DEGREE OF FREEDOM (DOF) ANALYSIS**

#### • DOF

- Number of variables that can be specified independently
- $-N_F = N_V N_E$ 
  - N<sub>F</sub> : Degree of freedom (no. of independent variables)
  - N<sub>V</sub> : Number of variables
  - N<sub>E</sub> : Number of equations (no. of dependent variables)
  - Assume no equation can be obtained by a combination of other equations

#### Solution depending on DOF

- If N<sub>F</sub> = 0, the system is *exactly determined*. Unique solution exists.
- If N<sub>F</sub> > 0, the system is *underdetermined*. Infinitely many solutions exist.
- If  $N_F < 0$ , the system is *overdetermined*. No solutions exist.

CHBE320 Process Dynamics and Control

Korea University 4-7

### LINEAR VERSUS NONLINEAR MODELS

### Superposition principle

 $\forall \alpha, \beta \in \Re$ , and for a linear operator, *L* Then  $L(\alpha x_1(t) + \beta x_2(t)) = \alpha L(x_1(t)) + \beta L(x_2(t))$ 

· Linear dynamic model: superposition principle holds

 $\begin{aligned} \forall \alpha, \beta \in \Re, u_1(t) \to y_1(t) \text{ and } u_2(t) \to y_2(t) \\ \alpha u_1(t) + \beta u_2(t) \to \alpha y_1(t) + \beta y_2(t) \end{aligned}$ 

$$\forall \alpha, \beta \in \Re, x_1(0) \to y_1(t) \text{ and } x_2(0) \to y_2(t)$$
$$\alpha x_1(0) + \beta x_2(0) \to \alpha y_1(t) + \beta y_2(t)$$

- Easy to solve and analytical solution exists.

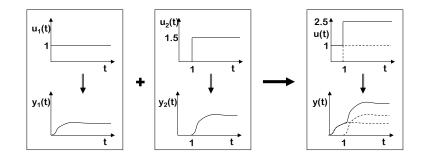
- Usually, locally valid around the operating condition

Nonlinear: "Not linear"

- Usually, hard to solve and analytical solution does not exist.

CHBE320 Process Dynamics and Control

# ILLUSTRATION OF SUPERPOSITION PRINCIPLE



- Valid only for linear process
  - For example, if y(t)=u(t)<sup>2</sup>,
    (u<sub>1</sub>(t)+1.5u<sub>2</sub>(t))<sup>2</sup> is not same as u<sub>1</sub>(t)<sup>2</sup>+1.5u<sub>2</sub>(t)<sup>2</sup>.

CHBE320 Process Dynamics and Control

Korea University 4-9

# **TYPICAL LINEAR DYNAMIC MODEL**

• Linear ODE

$$\tau \frac{dy(t)}{dt} = -y(t) + Ku(t) \quad (\tau \text{ and K are contant, 1st order})$$
$$\frac{d^{n}y(t)}{dt^{n}} + a_{n-1}\frac{d^{n-1}y(t)}{dt^{n-1}} + \dots + a_{0}y(t)$$
$$= b_{m}\frac{d^{m}u(t)}{dt^{m}} + b_{m-1}\frac{d^{m-1}u(t)}{dt^{m-1}} + \dots + b_{0}u(t) \quad (\text{nth order})$$

Nonlinear ODE

$$\tau \frac{dy(t)}{dt} = -y(t)^2 + Ku(t) \qquad \qquad \tau \frac{dy(t)}{dt}y(t) = -y(t)\sin(y) + Ku(t)$$
  
$$\tau \frac{dy(t)}{dt} = -y(t) + K\sqrt{u(t)} \qquad \qquad \tau \frac{dy(t)}{dt} = -e^{-y(t)} + Ku(t)$$

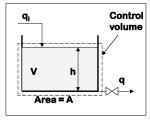
CHBE320 Process Dynamics and Control

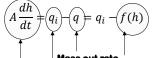
# MODELS OF REPRESENTATIVE PROCESSES

### Liquid storage systems

- System boundary: storage tank
- Mass in: q<sub>i</sub> (vol. flow, indep. var)
- Mass out: q (vol, flow, dep. var)
- No generation or disappearance (no reaction or leakage)
- No energy balance
- **DOF=2**  $(h, q_i)$  1=1
- If  $f(h) = h/R_V$ , the ODE is linear. ( $R_V$  is the resistance to flow)
- If  $f(h) = C_V \sqrt{\rho g h/g_c}$ , the ODE is nonlinear. ( $C_V$  is the valve constant)

CHBE320 Process Dynamics and Control





Mass out rate Outlet flow is a Mass in rate Accumulation rate in tank function of head

Korea University 4-11

- Continuous Stirred Tank Reactor (CSTR)
  - Liquid level is constant (No acc. in tank)
  - Constant density, perfect mixing
  - Reaction:  $\mathbf{A} \rightarrow \mathbf{B}$   $(r = k_0 \exp(-E/RT)c_A)$
  - System boundary: CSTR tank
  - Component mass balance

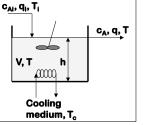
$$V\frac{dc_A}{dt} = q(c_{Ai} - c_A) - Vkc_A$$

- Energy balance

$$V\rho C_p \frac{dT}{dt} = q\rho C_p (T_i - T) + (-\Delta H)Vkc_A + UA(T_c - T)$$

- DOF analysis
  - No. of variables: 6  $(q, c_A, c_{Ai}, T_i, T, T_c)$
  - No. of equation:2 (two dependent vars.:  $c_A$ , T)
  - DOF=6 2 = 4
  - Independent variables: 4 ( $q, c_{Ai}, T_i, T_c$ )
  - Parameters: kinetic parameters, V, U, A, other physical properties
  - Disturbances: any of q,  $c_{Ai}$ ,  $T_i$ ,  $T_c$ , which are not manipulatable

CHBE320 Process Dynamics and Control



### **STANDARD FORM OF MODELS**

#### From the previous example

$$\frac{dc_A}{dt} = \frac{q}{V}(c_{Ai} - c_A) - kc_A = f_1(c_A, T, q, c_{Ai})$$
$$\frac{dT}{dt} = \frac{q}{V}(T_i - T) + \frac{q}{\rho C_p}(-\Delta H)kc_A + \frac{UA}{\rho C_p}(T_c - T) = f_2(c_A, T, q, T_c, T_i)$$

#### State-space model

 $\dot{\mathbf{x}} = d\mathbf{x}/dt = \mathbf{f}(\mathbf{x}, \mathbf{u}, \mathbf{d})$ where  $\mathbf{x} = [x_1, \dots, x_n]^T$ ,  $\mathbf{u} = [u_1, \dots, u_m]^T$ ,  $\mathbf{d} = [d_1, \dots, d_l]^T$ 

- **x: states,**  $[c_A T]^T$
- u: inputs,  $[q T_c]^T$
- **d: disturbances,**  $[c_{Ai} T_i]^T$
- y: outputs can be a function of above, y=g(x,d,u),  $[c_A T]^T$
- If higher order derivatives exist, convert them to 1<sup>st</sup> order.

CHBE320 Process Dynamics and Control

Korea University 4-13

# **CONVERT TO 1<sup>ST</sup>-ORDER ODE**

### Higher order ODE

 $\frac{d^n x(t)}{dt^n} + a_{n-1} \frac{d^{n-1} x(t)}{dt^{n-1}} + \dots + a_0 x(t) = b_0 u(t)$ 

### Define new states

 $x_1 = x, x_2 = \dot{x}, x_3 = \ddot{x}, \cdots, x_n = x^{(n-1)}$ 

• A set of 1<sup>st</sup>-order ODE's

 $\begin{array}{l} \dot{x}_1 = x_2 \\ \dot{x}_2 = x_3 \\ \vdots \\ \dot{x}_n = -a_{n-1}x_n - a_{n-2}x_{n-1} - \cdots - a_0x_1 + b_0u \end{array}$ 

CHBE320 Process Dynamics and Control

### SOLUTION OF MODELS

### ODE (state-space model)

- Linear case: find the analytical solution via Laplace transform, or other methods.
- Nonlinear case: analytical solution usually does not exist.
  - Use a numerical integration, such as <u>*RK method*</u>, by defining initial condition, time behavior of input/disturbance
  - Linearize around the operating condition and find the analytical solution
- PDE
  - Convert to ODE by discretization of spatial variables using *finite difference approximation* and etc.

$$\frac{\partial T_L}{\partial t} = -v \frac{\partial T_L}{\partial z} + \frac{1}{\tau_{HL}} (T_w - T_L) \longrightarrow \frac{dT_L(j)}{dt} = -\frac{v}{\Delta z} T_L(j-1) - \left(\frac{v}{\Delta z} + \frac{1}{\tau_{HL}}\right) T_L(j) + \frac{1}{\tau_{HL}} T_w \ (j = 1, \dots N)$$
$$\frac{\partial T_L}{\partial z} \approx \frac{T_L(j) - T_L(j-1)}{\Delta z}$$

CHBE320 Process Dynamics and Control

Korea University 4-15

## LINEARIZATION

### • Equilibrium (Steady state)

- Set the derivatives as zero:  $0 = f(\bar{x}, \bar{u}, \bar{d})$
- Overbar denotes the steady-state value and  $(\tilde{x}, \tilde{u}, \tilde{d})$  is the equilibrium point. (could be multiple)
- Solve them analytically or numerically using *Newton method*, etc.

#### Linearization around equilibrium point

- Taylor series expansion to 1<sup>st</sup> order

$$f(x,u) = f(\bar{x},\bar{u}) + \frac{\partial f}{\partial x}\Big|_{(\bar{x},\bar{u})} (x-\bar{x}) + \frac{\partial f}{\partial u}\Big|_{(\bar{x},\bar{u})} (u-\bar{u}) + \cdots$$

- Ignore higher order terms
- Define deviation variables:  $x' = x \bar{x}$ ,  $u' = u \bar{u}$

$$\dot{\mathbf{x}}' = \frac{\partial \mathbf{f}}{\partial \mathbf{x}} \bigg|_{(\bar{\mathbf{x}}, \bar{\mathbf{u}})} \mathbf{x}' + \frac{\partial \mathbf{f}}{\partial \mathbf{u}} \bigg|_{(\bar{\mathbf{x}}, \bar{\mathbf{u}})} \mathbf{u}' = \mathbf{A}\mathbf{x}' + \mathbf{B}\mathbf{u}'$$

**CHBE320 Process Dynamics and Control** 

